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Holt Chemistry Chapter 6. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. eman6044. The questions that she said will be on the test. Terms in this set (22) ... How does a covalent bond differ from an ionic bond? Ionic- Electrons are transferred Covalent- Electrons are shared.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

### 6 Chemical Bonding

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6) purely ionic. b. partially ionic. c. polar-covalent. d. nonpolar-covalent. 10. You can estimate the degree to which a bond between two atoms is ionic or covalent by calculating the a. distance between the atoms' nuclei. b difference in the atoms' electronegativities. c. atoms' atomic radii. d. number of atoms in the compound.

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