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Chemical Bonding

# Chapter 6

# Review

# Chemical

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## Chemical Bonding

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## Chemical Bonding Worksheet

### **Chapter 6 Review Chemical Bonding**

#### CHAPTER 6 REVIEW

#### Chemical Bonding

#### SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

- a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b)

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inner electrons (d)

Lewis structures 2. b A covalent bond consists of (a) a shared electron.

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## Chemical Bonding

chemical bond between atoms results from the attraction between the valence electrons and \_\_\_\_\_ of different atoms. nuclei. A covalent bond consists of.

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A chemical bond is the force that holds atoms or ions together as a unit. An ionic bond is

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## Chemical Bonding

the force that holds cations and anions together. An ionic bond forms when electrons are transferred from one atom to another. Formation of Ionic Bonds. ... Chapter 6 - Chemical Bonds

### **Chapter 6 - Chemical Bonds**

CHAPTER 6 REVIEW .

Chemical Bonding .

SHORT ANSWER

Answer the following questions in the space

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provided. 1. \_\_\_\_\_; b\_\_\_\_\_ In

metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2.

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Review Previous  
Concepts Chemical  
Bonding CHAPTER 6



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Section 1 Introduction  
to Chemical Bonding ...

SECTION 6.1 REVIEW

168 CHAPTER 6 (a)

Water molecule, H 2O

C 12H 22O 11 (c)

Sucrose molecule, (b)

Oxygen molecule, O 2

Most of the chemicals  
inside living things and  
produced by

## **CHAPTER 6 Chemical Bonding**

Modern Chemistry 45

Chemical Bonding

CHAPTER 6 REVIEW

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## Chemical Bonding

### SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1.

\_\_\_\_\_ The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule.

(d) atom. 2. \_\_\_\_\_ In a crystal of an ionic compound, each cation is surrounded by a ...

## **CHAPTER 6 REVIEW**

## **Chemical Bonding**

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## CHAPTER 6 REVIEW

### Chemical Bonding

#### SECTION 6-2 SHORT

ANSWER Answer the following questions in the space provided. 1.

Use the concept of potential energy to describe how a covalent bond forms between two atoms. 2.

Name two elements that form compounds that are exceptions to the octet rule.

## **6 Chemical Bonding**

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Chapter 6 review  
chemical bonding  
section 1 short answer  
answer the following  
questions in the space  
provided. Chapter 6  
chemical bonding in  
your computer by  
clicking resolution  
image. A a chemical  
bond between atoms  
results from the  
attraction between the  
valence electrons and  
of different atoms. B a  
covalent bond consists

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## Chemical Bonding Worksheet

of a a shared electron.

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#### **Answers -**

#### **Nidecmege**

#### CHAPTER 6 REVIEW

#### Chemical Bonding

#### SECTION 2 SHORT

ANSWER Answer the following questions in the space provided. 1.

Use the concept of potential energy to describe how a covalent bond forms between two atoms. As

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the atoms involved in  
the formation of a  
covalent bond  
approach each other,  
the

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emma\_pear. Terms in  
this set (32) Identify  
and define the three

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## Chemical Bonding

major types of chemical bonding. ... A covalent bond is a chemical bond that involves the sharing of electron pairs between atoms.

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Chemistry ... Chemical

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## Chemical Bonding

**Bond:** an attraction between two atoms resulting from a sharing of outer shell electrons or the presence of opposite charges on the atoms.

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CHAPTER 6 REVIEW  
Chemical Bonding  
SECTION 6-3 SHORT  
ANSWER Answer the  
following questions in



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the space provided. 1.

The notation for sodium chloride, NaCl, stands for one . (a) formula unit (c) crystal (b) molecule (d) atom

2. In a crystal of an ionic compound, each cation is surrounded by a number of .

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Chemical Bonding  
Worksheet  
Answers

Answer Key for each of  
the following types of  
bonds: a. ionic bond b.  
covalent bond c.

metallic bond 2. Use  
the electronegativity  
values shown in Figure  
5-20, on page 151 of  
the text, to determine  
HOMEWORK 6-1

Organic chemistry is  
study of carbon

## **Chapter 6 Review Chemical Bonding Answer Key**

This chapter 6 mixed  
*Page 18/27*

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should Helium have  
around its Lewis dot  
model? Chapter 6  
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years ago.

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chemical bonding

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Chemical Bonding

SECTION 1 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

a A chemical bond

between atoms results

from the attraction

between the valence

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## Chemical Bonding

electrons and of  
different atoms. (a)  
nuclei (c) isotopes (b)  
inner electrons (d)  
Lewis structures 2. b A  
covalent

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Bonds. ... Are the atoms more stable when they are bonded together or when they are apart? Chapter 6 Review Game DRAFT. 9th grade. 54 times. Chemistry. 68% average accuracy. 7 months ago.

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Hybridization

delocalizes the bonding electrons over the surface of the entire molecule while molecular orbital theory accounts for bonds one at a time.

Hybridization simplifies molecular orbital theory by only considering one bond at a time. Molecular orbital theory for

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polyatomic molecules  
considers all of the  
bonds at once.

## **Review of Chemical Bonding: Review Test | SparkNotes**

In Chapter 6, we will begin studying how atoms interact with each other to form chemical bonds. Students will review the differences between ionic and covalent bonding and will learn to recognize

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## Chemical Bonding

examples of each,  
including how to  
calculate ionic  
character using  
electronegativity  
values.

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