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(s) A)exothermic, released B)exothermic, absorbed.

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Outline 5.1 The Nature of Energy

Thermodynamics is the study of energy and its transformations.

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Thermochemistry is the study of the relationships between chemical reactions and energy changes. Kinetic Energy and Potential Energy Kinetic energy is the energy of motion: $E = \frac{1}{2}mv^2$ Potential energy is the energy an object possesses by virtue of its position.

AP Chemistry: Thermochemistry

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Lecture Outline

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Multiple Choice and FIB (40%) 1) A chemistry student is filtering and drying a precipitate that formed from two solutions reacting. Which one is the most likely reason for an exceptionally high % yield? A) Careless filtering and sufficient drying B) Careless filtering and

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insufficient drying

AP Chemistry Test (Chapter 3)

AP Chemistry - Agenda for January 2016

Ch. 6 Electronic Structure of Atoms &

Chapter 7 Periodic Properties of the

Elements ... Finish chapter 6 Electronic

Structure. Check Book questions on

quantum numbers ... Self Test Questions

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(7.1-7.10, 7.19-7.21, 7.29 & 7.30, 7.36-7.40, 7.43-7.58) to be done in class ...

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Chapter 8 - Lewis Structures.

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% b. 6.6 % c. 3.3 % d. 5.0×10^{-4} % e. 2.2
% 2. A quantitative observation a.
contains a number and a unit. ...

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AP Chemistry Test (Chapter 2) #1

Name _____ Please give the formula for
each one. 1) _____ Vanadium (V)

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thiosulfate

AP Chemistry Test (Chapter 2) #1 Name

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AND EXPLANATIONS. Multiple-Choice. 1. A The equivalence point is defined as the point at which the moles of acid is equal to the moles of base. The moles of NaOH ...

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